



SYNTHESIS AND CHARACTERIZATION OF SOME BENZOTHIAZOLOPYRAZOLINE DERIVATIVES

Veena Pareek^{*1}, Pradeep K. Paliwal², Shubha Jain³

^{*1}Department of Chemistry, Government College Nagda, Vikram University, Ujjain (MP), 456010, India

²Nirmala College of Education, Vikram University, Ujjain (MP), 456010 India

³School of Studies in Chemistry & Biochemistry, Vikram University, Ujjain (MP), 456010, India



Abstract:

A new route for the synthesis of benzothiazolopyrazoline derivatives has been developed using various chalcone derivatives and 2-hydrazinobenzothiazole and piperidine catalyst. The product obtained in shorter reaction times and piperidine behaves as good catalyst for the cycloaddition of chalcone and 2-hydrazinobenzothiazole. The reaction carried out in aqueous-ethanol medium at reflux condition and product obtained in high yield.

Keywords: Benzothiazolopyrazoline; Chalcones; Cycloaddition Reactions; Piperidine.

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1. Introduction

Heterocycles containing nitrogen, sulphur and oxygen constitute the core structure of a number of biologically interesting compounds. Pyrazolines are an interesting group of compounds, many of which possess pharmacological properties such as analgesic, antidepressant, antipyretic and antirheumatic activities, anti-inflammatory activity and are used as potent antidiabetic agents. In recent times, pyrazolines were reported as a DP-IV inhibitors and antitumor agents⁶⁻⁸. Some nitrogen heterocyclic compounds were reported to be anticancer⁹, antimicrobial¹⁰⁻¹⁴ and anti-inflammatory agents¹⁵. Because of interesting biological activity of various pyrazoline derivatives has been focused on this group. In addition, pyrazolines have played a crucial part in heterocyclic chemistry and also used broadly in organic synthesis¹⁶⁻²⁰. Benzothiazole and its derivatives²¹⁻²⁵ have been known as a class of medicinal importance. Benzothiazole derivatives represent an extensive group of heterocyclic compounds, several of which have already found application in the medical sphere as medicines²⁶ as well as in agriculture²⁷. Many of substituted benzothiazoles are known as substances with antibacterial and antifungal activities²⁸⁻³⁰ and are reported also to be active as antineoplastics³¹ agent. Keeping in view the biological and clinical importance of benzothiazoles and pyrazolines, in this paper we report the synthesis and biological activity of new 3,5-diaryl-1-benzothiazolopyrazoline derivatives.

2. Experimental

2.1. General Procedures

Melting points were determined by the open capillary method. The purity of the compounds was controlled by thin layer chromatography (TLC). IR spectra were recorded as KBr pellets on Perkin-Elmer spectrum spectrophotometer. Carbon, hydrogen, nitrogen and sulphur were estimated by elemental analysis. ¹H-NMR spectra were recorded on Waters spectrometer. Mass spectra were measured on JEOL SX mass spectrometer. Substituted chalcones (**1a-d**) were prepared according to the reported methods.³²

2.2. General Procedure for the Preparation of Compounds

A mixture of substituted chalcone (**1a-d**, 10mmol) and 2-hydrazinobenzothiazole (10mmol) in EtOH-H₂O (20 mL) and 2drops of piperidine is added as catalyst then reaction mixture was refluxed for 30 min. and then cooled to room temperature. The precipitate was separated by filtration, washed with water and recrystallized with ethanol to obtained 3, 5-diaryl-1-benzothiazolopyrazoline derivatives.

3, 5-Diphenyl-1-Benzothiazolo Pyrazoline

Obtained as white crystals yield 94%, m.p. 175-176°C; ¹H NMR (δ): 3.24 (1H CH₂(Pyraz)), 3.90 (1H, CH₂(Pyraz)), 5.80 (1H, CH(Pyraz)), 6.90-7.68(14H, m, Ar-H); IR(ν) max: 3428, 3016, 1596, 1495, 1443, 1390, 1280, 1120, cm⁻¹; MS, *m/z*: 355, 354, 278, 233, 154, 126, 112, 100, 86. Anal. Calcd. For C₂₂H₁₇N₃S: C, 74.36; H, 4.78; N, 11.83; S, 9.02. Found: C, 74.44; H, 4.68; N, 11.88; S, 9.07%.

3. Results and Discussion

Formation of 3, 5-diarylpyrazoline derivatives by the reaction of α,β -unsaturated ketones and hydrazines may take place under various reaction conditions using aqueous-ethanol as solvent. In our present case to obtained 3, 5-diaryl-1-benzothiazolopyrazoline derivatives (**3a-d**) a mixture of substituted chalcone (**1a-d**, 10mmol) and 2-hydrazinobenzothiazole (10mmol) in ethanol-water 1:1 (20 mL) was refluxed for 30 min. and then cooled to room temperature. The precipitate was separated by filtration, washed with water and recrystallized from alcohol (Scheme 1). Substituted chalcones (**1a-d**) were prepared according to the reported methods. Structures of all new compounds (**3a-d**) have been elucidated by elemental analyses, ¹H NMR, Mass spectral and IR measurements. In the IR spectra of 3,5-diaryl-1-benzothiazolopyrazoline derivatives (**3a-d**) there is no carbonyl absorption but absorption bands in the region 1600-1450 cm⁻¹ due to the presence of C=C and C=N stretching vibrations. Stretching vibrations due to the intramolecular hydrogen bonding of -OH group gave absorption in the region 3000-2500cm⁻¹. In addition to aromatic protons, the ¹H NMR spectra of these compounds exhibit double doublets for each δ CH₂ proton between δ 3.23 to 3.64 and δ 3.88 to 4.00 and double doublet between δ 5.25 to 6.22 for -CH proton of the pyrazoline nucleus system.

3a: R1= R2 = R3= R4= R5= H

3b: R2= OH, R1= R3= R4= R5= H

3c: R1= Cl, R2 = R3= R4= R5=H

3d: R1= CH₃, R2 = R3= R4= R5= H

In conclusion, we have synthesized a series of new 3, 5-diaryl-1-benzothiazolopyrazoline derivatives. These substituted benzothiazolo- pyrazolines are very stable compounds, which are expected to biological active against various bacterial and fungal stains.

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*Corresponding author.

E-mail address: pallavipareek.pareek7@ gmail.com/ paliwalchemi@ gmail.com